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[Enthalpy of vaporization - Wikipedia](#)

In relations to the first law of thermodynamics, we can see that by adding heat ( $Q$ ) or work ( $W$ ) the internal energy of the gaseous system can be increased. Also, that during compression of the system, the volume of the gas will decrease and response its temperature will increase and thus the internal energy of the system will also increase since temperature is related to energy. And this is

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**This means that solutions to mass balance problems that make this assumption must be taken as approximations to reality. If completely mixed systems exist, or at least systems that we can approximate as completely mixed, then it stands to reason that some systems are completely unmixed or approximately so. These systems are called plug-flow systems. The behavior of a plugflow system is ...**

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