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The history of chemistry represents a time span from ancient history to the present. By 1000 BC, civilizations used technologies that would eventually form the basis of the various branches of chemistry. Examples include the discovery of fire, extracting metals from ores, making pottery and glazes, fermenting beer and wine, extracting chemicals from plants for medicine and perfume, rendering ...

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Approaches chemistry as the study of elements and compounds without the later emphasis on bond theory and analysis. ... Physical chemistry Physical Chemistry (Atkins and de Paula) P. W. Atkins; Oxford University Press, 1st Ed. 1978, 10th Ed. 2014 (with Julio de Paula from 7th Ed. 2002) Description: A classic general textbook for an undergraduate course in physical chemistry Importance: This ...

[physical chemistry - Calculating Compressibility factor ...](#)

The standard-state free energy of formation is the change in free energy that occurs when a compound is formed from its elements in their most thermodynamically stable states at standard-state conditions. In other words, it is the difference between the free energy of a substance and the free energies of its constituent elements at standard-state conditions: $\Delta G^{\circ} = \sum \Delta G^{\circ}_{f \dots}$